

hw11 , due: Friday, December 6 at 4pm

1. Consider the power series $\sum_{n=0}^{\infty} \frac{(x-1)^n}{2^n}$. Find the interval of convergence; find the function $f(x)$ represented by the series; sketch the graph of $f(x)$ for $-3 \leq x \leq 7$; indicate the center of the series and interval of convergence on the x -axis; indicate any asymptotes with a dashed line.

2. Find $T_1(x)$ and $T_3(x)$, the 1st and 3rd degree Taylor polynomials for $f(x) = \sin x$ at $a = \pi$. Sketch the graphs of $f(x), T_1(x), T_3(x)$ on the same plot for $-2\pi \leq x \leq 2\pi$ and label each one.

3. Express $\int_0^{1/2} \frac{dx}{1+x^6}$ as a series; give the first 3 terms and the general term. How many terms are needed in the partial sum to ensure the error is less than 10^{-5} ? Use an error bound derived in class.

4. We can approximate $\sqrt{10}$ using the fact that it is close to $\sqrt{9} = 3$.

(a) Find $T_1(x)$, the 1st degree Taylor approximation for $f(x) = \sqrt{x}$ at $a = 9$, and evaluate $T_1(10)$.

(b) Repeat for $T_2(x)$, the 2nd degree Taylor approximation.

Express the answers as a rational number (i.e. m/n , where m, n are integers) and in decimal form with eight decimal digits. Then find the exact value of $\sqrt{10}$ using a calculator and compute the error in each case; does the Taylor approximation become more accurate as the degree increases?

5. The Arrhenius equation, $k = Ae^{-E/RT}$, gives the rate constant k of a chemical reaction in terms of the temperature T , where A is a constant, E is the activation energy, and R is the universal gas constant (these are all positive). Find $\lim_{T \rightarrow 0} k, \lim_{T \rightarrow \infty} k$. Sketch the graph of k versus T for $T \geq 0$ (hint: see page 66 in the notes). As the temperature increases, does the reaction occur slower or faster?

6. A non-reflecting object that absorbs all incident radiation is called a blackbody and in thermal equilibrium it emits radiation with energy density $f(\lambda)$, where λ is the radiation wavelength. The Rayleigh-Jeans law for energy density is $f_{RJ}(\lambda) = 8\pi k_B T \lambda^{-4}$, where k_B is Boltzmann's constant and T is the object's temperature; it agrees with experiments for long wavelength, $\lim_{\lambda \rightarrow \infty} f_{RJ}(\lambda) = 0$, but not for short wavelength; experiments show that $\lim_{\lambda \rightarrow 0} f(\lambda) = 0$, but $\lim_{\lambda \rightarrow 0} f_{RJ}(\lambda) = \infty$; this is called the ultraviolet catastrophe. In 1900 Max Planck used the novel idea of energy quantization to derive a

different expression, $f_P(\lambda) = \frac{8\pi hc \lambda^{-5}}{e^{hc/\lambda k_B T} - 1}$, where h is Planck's constant and c is the speed of light.

a) Show that $\lim_{\lambda \rightarrow \infty} f_P(\lambda) = 0$ and $\lim_{\lambda \rightarrow 0} f_P(\lambda) = 0$; hence Planck's law agrees with experiments in the long and short wavelength limits (hint: set $x = hc/\lambda k_B T$).

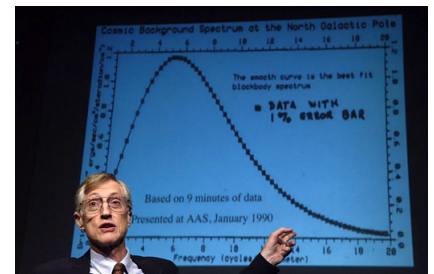
b) Show that $f'_P(0) = 0$ (hint: use the definition, see page 66 in the notes)

c) Show that $f_P(\lambda) \sim f_{RJ}(\lambda)$ for $\lambda \rightarrow \infty$; recall that this means $\lim_{\lambda \rightarrow \infty} f_P(\lambda)/f_{RJ}(\lambda) = 1$.

d) Show that $f_P(\lambda) < f_{RJ}(\lambda)$ for $\lambda \geq 0$.

e) Sketch $f_P(\lambda), f_{RJ}(\lambda)$ for $\lambda \geq 0$ on the same plot and label them.

Planck's discovery is described in the article, "Max Planck: the reluctant revolutionary"; see link on Canvas site. In 2006 John Mather and George Smoot won the Nobel Prize in Physics, "for their discovery of the blackbody form ... of the cosmic microwave background radiation"; the photo shows Mather with a plot of the experimental data in excellent agreement with Planck's law.



announcement The final exam is on Thursday, December 12 at 8-10am (sections 1, 2, 5 in 1200 CHEM, sections 3, 4 in 1300 CHEM). It covers the entire course; calculators are not allowed; we will supply the exam booklets; the exam will be uploaded to Gradescope; you may use 2 sheets of notes (e.g. 2 sides of one page).